Y12 – Alkenes Checklist

this module you are expected to be able to	In Notes	Revised
Define the terms		
π-bond		
 Electrophile 		
 Stereoisomer 		
o E/Z isomer		
 Cis/Trans isomer 		
Recall the bond angle and shape of ethene and therefore alkenes in general.		
• Explain how a π -bond is formed and why it is more reactive than a σ -bond.		
Name alkenes using IUPAC nomenclature.		
Draw a reaction mechanism for electrophilic addition.		
 Predict the reactions of alkenes with hydrogen, water, halogens and hydrogen halides including all reagents and conditions. 		
Predict which structural isomer will form from an asymmetrical alkene using Markownikoff's rules.		
Use Cahn-Ingold-Prelog nomenclature to assign E/Z isomerism.		
Draw stereoisomers of alkenes given different reaction schemes and structural formulae.		
 Describe addition polymerisation using equations, and be able to predict monomer/polymers from the reaction scheme. 		
Give benefits of combustion and feedstock recycling, and describe how HCl is removed from waste gases when burning PVC.		
Give benefits of biodegradable and photodegradable polymers.		

I have						
Updated my yellow book notes						
Ensured I understand all of my notes						
Looked on the open drive for additional work						
Asked my teacher for guidance						
Confidence rating	I'm doomed!	 -	=	+	++	I am the BOSS!