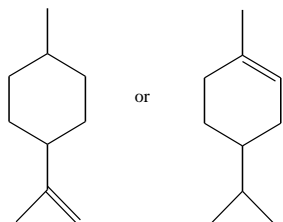


## ALKENES MS

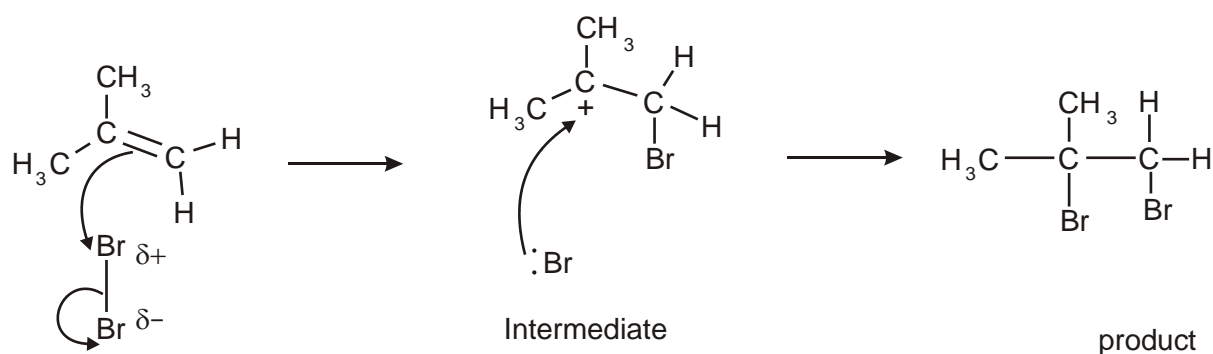
1. (a) (i)  $C_5H_8$  1  
 (ii)  $C_5H_8$  1  
 (b) (i) Ni/Pt/Pd 1  
 (ii) 1 mark for  $C_5H_{12}$  1  
 1 mark for correct balancing 1  
 (iii)



1  
**[6]**

2. (a) (i) alkene 1  
 bromine 1  
 decolourises 1  
 (ii) 3-methylhex-2-en-1-ol/ 1-hydroxy-3-methylhex-2-ene 1  
**[4]**

3. (i) *electrophile*: lone pair (of electrons) acceptor. ✓ 1  
 (ii)



essential mark intermediate carbocation/carbocation ion, accept primary  
 /"triangular"/ ✓

essential mark product ✓

curly arrow from double bond to  $Br_2$  ✓

curly arrow showing movement of electrons in the Br-Br bond **or** the dipole in  
 the Br-Br ✓

curly arrow from lone pair of electrons in  $Br^-$  to intermediate ✓

*mark any errors first*

5 max

**[5]**

- |           |                                       |   |
|-----------|---------------------------------------|---|
| <b>4.</b> | $\text{CH}_3\text{CBr}_2\text{CH}_3$  | 1 |
|           | $\text{CH}_3\text{CHBrCH}_2\text{Br}$ | 1 |
|           | $\text{CH}_3\text{CH}_2\text{CHBr}_2$ | 1 |

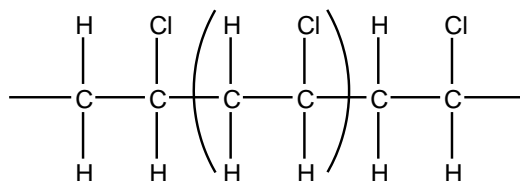
( $\text{CH}_3\text{CHBrCH}_2\text{Br}$  has a chiral centre, hence optical isomers of 1, 2-dibromopropane are acceptable but must be drawn with 'wedge-shape' bonds and be non-superimposable mirror images)

**[3]**

- |           |      |   |   |
|-----------|------|---|---|
| <b>5.</b> | (i)  | $\text{C}_9\text{H}_{20} \longrightarrow \text{C}_7\text{H}_{16} + \text{C}_2\text{H}_4$  | 1 |
|           | (ii) | $\text{C}_2\text{H}_4 + \text{H}_2\text{O} \longrightarrow \text{C}_2\text{H}_5\text{OH}$ | 1 |
|           |      | temperature > 100 °C/ steam   | 1 |
|           |      | phosphoric acid (catalyst)  | 1 |

**[4]**

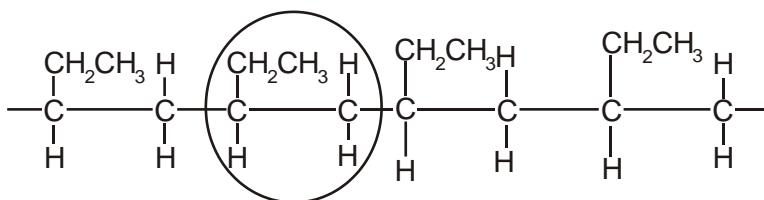
6. (i) correctly shows three repeat units with 'end bonds' 1  
correctly identifies the repeat unit 1



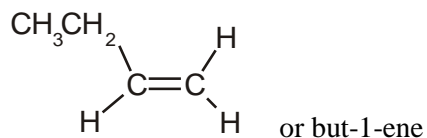
- (ii) harmful/toxic fumes are produced 1  
(iii) recycle/remove HCl by using gas scrubbers or wtte/crack polymers/used a feedstock/ source of fuel (in an incinerator)/developing biodegradable alternatives. 2

[5]

7. (i) Addition (not additional) ✓ 1  
(ii) ✓ 1



- (iii) ✓ 1



- (iv) Poly(but-1-ene) ✓ 1

[4]

**8. 1st bullet**

product:  $\text{CH}_3\text{CH}_2\text{CHBrCH}_2\text{Br}$  (1)

equation:  $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2 + \text{Br}_2 \rightarrow \text{CH}_3\text{CH}_2\text{CHBrCH}_2\text{Br}$  (1)

products:  $\text{CH}_3\text{CH}_2\text{CHBrCH}_3$  **and**  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Br}$  (1)

(or statement that 2-bromo- is formed)

equation:  $\text{CH}_3\text{CH}=\text{CHCH}_3 + \text{HBr} \rightarrow \text{CH}_3\text{CH}_2\text{CHBrCH}_3$  (1)

(i.e. for one product)

products:  $\text{CH}_3\text{CH}_2\text{CHOHCH}_3$  **and**  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$  (1)

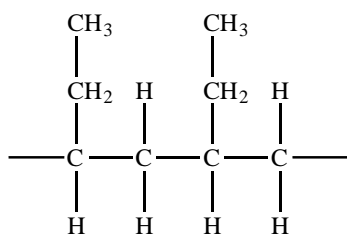
(or statement that 2-ol is formed)

equation:  $\text{CH}_3\text{CH}=\text{CHCH}_3 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{CH}_2\text{CHOHCH}_3$  (1)

(i.e. for one product)

6

**2nd bullet**



1 mark for skeleton with two repeat units (1)

1 mark for correct groups on side chains (1)

2

**3rd bullet**

two (1) (1) from

energy from incineration

development of biodegradable polymers

cracking of waste polymers

2

**[10]**