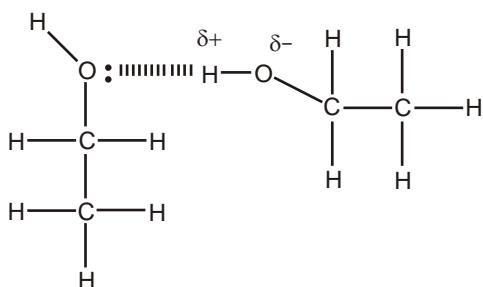


HALOGENOALKANES AND ALCOHOLS HW MS

1.



dipoles 1

hydrogen bond between O in one O-H
and H in the other O-H 1

lone pair from O involved in the H-bond 1

[3]

2. (a) (i) C_4H_{10} ✓ 1

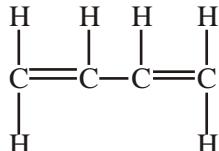
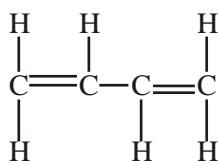
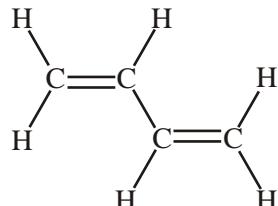
(ii) $\text{C}_2\text{H}_5\text{O}$ ✓ 1

(iii) B and E ✓ 1

(iv) A and F ✓ 1

(b) $(\text{C}_4\text{H}_9\text{OH} \rightarrow) \text{C}_4\text{H}_8 + \text{H}_2\text{O}$ ✓ 1

(c) any unambiguous formula: ✓ 1

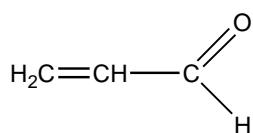


buta-1,3-diene ✓ 1

name ecf to the structure only if structure above has formula C_4H_6

[7]

3. (a) (i) prop-2-en-1-ol $\text{CH}_2=\text{CHCH}_2\text{OH}$ must show the C=C double bond
acrolein



must clearly show the aldehyde group and the C=C

1

- (ii) alkene/C=C double bond

1

- (b) (i) acidified / H^+
dichromate/ $\text{Cr}_2\text{O}_7^{2-}$

1

- (ii) $\text{CH}_2\text{CHCH}_2\text{OH}/ \text{C}_3\text{H}_6\text{O}/ \text{C}_3\text{H}_5\text{OH} + [\text{O}] \longrightarrow \text{CH}_2\text{CHCHO}/ \text{C}_3\text{H}_4\text{O}/$
 $\text{C}_2\text{H}_3\text{CHO} + \text{H}_2\text{O}$
not CH_2CHCOH

1

[6]

4. (a) (i) H_2SO_4 – any mention of (aq) loses the mark
(ii) any correct formula/structure or name for benzoic acid

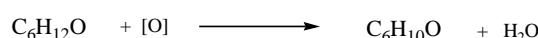
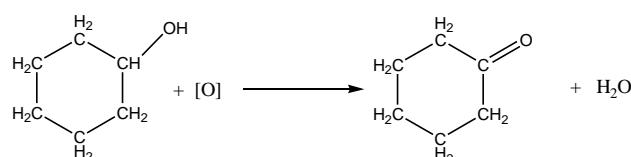
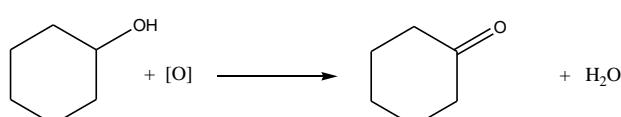
1

1

- (b) (i) dichromate/ $\text{Cr}_2\text{O}_7^{2-}$ /permanganate
(ii)

1

1



[4]